Carbon-based fibers are of practical importance in constructing high-strength, low-weight structural materials in the transportation vehicles, civil engineering and defense. Fabrication of carbon nanotube (CNT) fibers/yarns has been attracting tremendous attentions for years with the desire to integrate such unique properties as high strength, electrical and thermal conductivities of the individual nanotubes into the useful, macroscopic ensembles. Breakthroughs have been made in wet approaches[1–4] and dry-state spinning of CNTs[5–7] for CNT fiber fabrication. CNT is the typical one dimensional (1D) nanostructure with high aspect ratio and strong binding force between its bundles.[8] As a result, just like the spinning of cotton and wool, CNT fibers and sheets can be directly drawn out from their arrays[5,7,8] or high concentration suspensions.[2–4] However, beyond the superb properties of as-produced CNT fibers, we have to face the high cost for producing the initial superaligned CNT arrays,[5,7,8] and extremely rigorous conditions such as high temperature (＞1000 °C),[8] and caustic media (e.g., fuming sulfuric acid[9] and chlorosulphonic acid[5]) for CNT dispersion.

Graphene, a 2D monolayer of carbon atoms packed into a honeycomb lattice, is the basic building block for carbon materials of all other dimensionalities,[10] which gains immense recognition due to its fascinating properties such as giant electron mobility,[11–13] high thermal conductivity,[14] extraordinary elasticity and stiffness.[15] However, the lack of scalable assembly methods has been a key hurdle in manufacturing graphenes. Therefore, for integration of the remarkable properties of individual graphenes into advanced, macroscopic, functional structures for practical applications, an effective assembly strategy in a well-controlled way has to be developed. Owing to the ease of preparation and processing, as well as the scalable production and low cost, chemically derived graphene sheets hold the promise for integration into various macroscopic architectures. In virtue of the intrinsic 2D structure, graphenes have been conformably assembled into 2D macroscopic configurations such as papers,[16–18] transparent and conductive films[19,20] and even 3D frameworks.[21,22] However, it is of an extraordinary challenge to directly assemble 2D microcosmic graphene sheets into macroscopic fibers, because the irregular size and shape of chemically derived graphenes, and the movable layer-by-layer stacking of graphenes in contrary to the highly tangled CNT assembly could seriously obstruct the formation of graphene fibers and impair their macroscopic mechanical properties. As a consequence, up to date, few attempts have been successfully made to assemble graphenes into macroscopic fibers[23,24] although we are aware of the lower cost of graphene fibers than that of both CNT and commercial carbon fibers, and their practical importance for various applications.[25]

In this work, we developed a facile one-step dimensionally-confined hydrothermal strategy to fabricate the neat graphene fibers from cost-efficient aqueous graphite oxide (GO) suspensions. A glass pipeline of 0.4 mm in inner diameter was used as the reactor (Figure S1). The aqueous GO suspension of 8 mg/mL was injected into the glass pipeline, followed by baking it at 230 °C for 2 h after sealing up the two ends of the pipeline. Finally, a graphene fiber matching the pipe geometry was produced.

Figure 1a shows a 63 cm long graphene fiber with a uniform diameter of ~33 μm. 1 mL GO suspension (8 mg/mL) will generate more than 6 meter long graphene fiber. The fiber diameter and length can be controlled by either simply using the pipeline with predesigned length and inner diameter or adjusting the initial GO concentration (Figure S2 and S3). The initially formed graphene fiber released from pipeline in wet state has a relative large diameter as exemplified in Figure S4a and S5a, which has a noncompact but highly 3D cross-linking porous structure of random graphene sheets (Figure S5b,c), characteristic of the hydrothermally produced samples.[21] After being dried in air, the graphene fiber almost maintains the length (Figure S4) but its diameter has a five times reduction compared with the wet one because the water loss caused the shrinkage (Figure S5a and Movie S1). The drastic contraction in the diameter direction during the drying process could lead to the surface tension force for spontaneous orientation of graphene sheets to some extent. The capillary force during the loss of water could induce the close packing of porous graphene sheets. Consequently, the loose linkage of graphene sheets

Z. Dong, C. Jiang, H. Cheng, Y. Zhao, Prof. L. Qu
Key Laboratory of Cluster Science
Ministry of Education
School of Chemistry
Beijing Institute of Technology
Beijing 100081, China
E-mail: lqu@bit.edu.cn
Prof. G. Shi
Department of Chemistry
Tsinghua University
Beijing 100084, China
Prof. L. Jiang
Laser Micro-/Nano-Fabrication Laboratory
School of Mechanical Engineering
Beijing Institute of Technology
100081, China

[+] These authors contributed equally to this work.

DOI: 10.1002/adma.201200170
becomes densely stacked and relatively aligned directionally to the fiber’s main axis (Figures 1b–e and S5d&e). The broken section shows the individual graphene sheets and their compact entanglement (Figure 1e). Transmission electron microscopic (TEM) investigation and electron diffraction pattern confirm that the fiber is made out of graphene sheets (Figure S6). The alignment of graphene sheets was also evaluated by the polarized Raman spectroscopy.[2,26,27] The ratio of G-band intensity (parallel versus perpendicular) provides a useful probe of the relative degree of alignment. The neat graphene fibers possess a Raman ratio of about 6:1 (Figure S7), while the ratio is about 1:1 for the initial wet graphene fiber, indicating that the graphene
sheets are being aligned parallel to the axis of the dry fiber, which is in agreement with the aforementioned SEM observations. We have not observed any aggregation and preformed orientation of initial GO before and after being injected into the pipeline (Figure 58). Therefore, the alignment of graphene sheets directionally to the fiber’s main axis could be attributed to the effect of capillary-induced shear force and surface tension-induced tensile force during dry process. Just like the daily used threads, the graphene fiber can be curved into coils and enlaced in bundles in wet and dry states (Figures 1f,g). The fiber does not break as the knot is tightened (Figure 1h), and two-ply yarn can be obtained by twisting two fibers (Figure 1i). These observations demonstrate the flexibility and resistance to torsion of graphene fibers, which is similar to CNT fibers.\(^{3,[6,7]}\)

Although derived from micrometer-scale GO sheets (Figure 58), the hydrothermally converted graphene fiber has a measured tensile strength of up to 180 MPa (Figure 2a). This value is close to that of the singles yarns of multiwalled CNTs,\(^{7}\) but much better than those of single-walled CNT fibers fabricated by wet spinning process.\(^{2–4}\) Higher strength of about 420 MPa is observed for the graphene fibers thermally treated at 800 °C for 2 h in vacuum (Figure 2a). The dry graphene fibers have an average density of 0.23 g/cm\(^3\), which is about 7 times lower than that of carbon fibers (about 1.7–2.0 g/cm\(^3\)), and over 3 times lighter than any previously reported CNT yarns (0.8 g/cm\(^3\) in ref. [7], 1.1 g/cm\(^3\) in ref. [2], 1.3–1.5 g/cm\(^3\) in ref. [3], and 2.0 g/cm\(^3\) in ref. [6]). As a consequence, the density-normalized failure stress of the graphene fibers is up to 782 MPa/g cm\(^{-3}\), exceeding the polymer-free CNT yarns (50–575 MPa/g cm\(^{-3}\))\(^{2,3,6,7}\) The graphene fibers have typical elongations at break of about 3 to 6%, which is comparable to that of CNT fibers\(^{1,7}\) but much larger than graphite fibers (about 1%). The elastic behavior at room temperature before break of graphene fibers presumably arises from the possible displacement of the graphene sheets within the fibers.

The hydrothermally synthesized graphene fibers have a four-probe electrical conductivity of ~10 S/cm at room temperature, which is similar to those of wet-spun single-walled CNT fibers.\(^{3,4}\) We also investigated the effect of bending on the electrical resistance of graphene fiber. The bend tests were carried out with an in-house-made two-point bending device and a high-precision mechanical system. The electrical resistance remains stable and shows a negligible variation in either the bent or straight state over 1000 cycles (Figure 2b). The excellent electromechanical stability gives graphene fiber the great potential as flexible conductors for curvilinear large-area electronics such as sensory skins and wearable communication devices.\(^{18}\)

The flexible and mechanically stable graphene fibers can be shaped to specific geometry on demand. The wet graphene fibers facilitate the prefabrication into various configurations, which will remain the predesigned structure once being dried. As shown in Figure 3a, circle, triangle and quadrangle of graphene fibers have been achieved by wrapping the wet graphene fiber around the cylindrical, triangular and quadrangular bars. The graphene fiber can even be folded into 3D architecture as exemplified by a tetrahedron in Figure 3b. A graphene fiber spring with stretchable and compressible character is also fabricated (Figure 3c–g), which shows the mechanically stable shape-memory feature.

The pliability of graphene fibers facilitates the weaving into a variety of macroscopic objects. For example, a textile structure of graphene fibers with self-supported stability was demonstrated in Figure 3f,g, which displays the great potentials in electronic-textile applications in view of the good conductivity of graphene fibers. In addition, the graphene fibers can be expediently integrated into flexible, transparent, conductive composite film. Figure 3h shows a graphene fiber network embedded into the polydimethylsiloxane (PDMS) matrix with tolerance to deformation.

Integration of functional components into the fibers will facilitate the important applications such as stimulus-responsive and electronic-textiles. The graphene fibers provide a platform for in-situ and post-synthesis incorporation of various materials with unique properties into the fiber bodies for wearable multifunctional applications. As an example, magnetic graphene fibers have been fabricated by introduction of Fe\(_2\)O\(_4\),

---

**Figure 2.** Mechanical and electrical properties of graphene fibers. (a) Typical stress-strain curves of single as-prepared (blue square) and 800 °C thermally treated (red circle) graphene fiber. (b) Electrical-resistance change of a graphene fiber upon repeated bending to a radius of 2 mm and then straightening for over 1000 cycles. The inset photographs show the straight and bending status of a graphene fiber (scale bars = 1 cm).
Large scale preparation of graphene fibers can be done by scaling up the pipelines and ovens, and continuous production of graphene fibers by setup of a thermal-flow circulatory pipeline system is underway. The strong but low-weight graphene fibers with appropriate functionalization, as well as the controllable shaping and stitchability, could be integrated into flexible and hierarchically engineered structures for a variety of specific applications such as smart clothing and electronic textiles.

Experimental Section

For preparation of graphene fibers, 8 mg/mL aqueous GO suspension (weight ratio of Fe₃O₄/GO = 1:10), followed by hydrothermal process within pipelines for in-situ incorporation of Fe₃O₄ into the interlayers of graphene sheets (Figure S9). The Fe₃O₄/graphene fiber with good mechanical flexibility possesses a sensitive magnetic response. As demonstrated in Figure 4a,b and Movies S2,S3, the magnetized graphene fiber can iteratively bend to the magnet from the initial upright position. On the other hand, the initially formed graphene fiber as-released from the pipeline has a porous network structure (Figure S5b,c), which allows to spontaneously immobilize the embedded guest components during drying process. As demonstrated in Figure S10, the TiO₂ nanoparticles have been intercalated into the framework of graphene sheets by soaking the wet graphene fiber in commercial TiO₂ (AEROXIDE P25) aqueous suspension (10 mg/mL) for 20 min with slight stir. After drying and annealing at 400 °C for 30 min, the TiO₂/graphene fiber (~8 wt%) exhibits a fast photocurrent response with good repeatability (Figure 4c). The strong photocurrent response indicates a direct electron/hole injection between the TiO₂ particles and graphene sheets through photoexcitation of TiO₂,[29] suggesting the significant applications in optoelectronic systems such as photodetectors, photocatalysts and photovoltaic cells. In spite of the absence of optimizing the fabrication process in this primary study, Fe₃O₄- and TiO₂/graphene fibers are still flexible and have a half of the tensile strength of neat graphene fibers.

Figure 3. Shaping and weaving of graphene fibers. (a) and (b) The handmade planar and 3D geometric structures of graphene fibers, respectively. (c–e) The spring of graphene fiber at free, stretched, and compressed status, respectively. (f and g) Photographs of the hand-knitted textile of graphene fibers. (h) Photograph of graphene fiber network embedded in PDMS matrix.

nanoparticles into the fibers. The magnetic Fe₃O₄ nanoparticles were well mixed with GO suspension under ultrasonication (weight ratio of Fe₃O₄/GO = 1:10), followed by hydrothermal process within pipelines for in-situ incorporation of Fe₃O₄ into the interlayers of graphene sheets (Figure S9). The Fe₃O₄/graphene fiber with good mechanical flexibility possesses a sensitive magnetic response. As demonstrated in Figure 4a,b and Movies S2,S3, the magnetized graphene fiber can iteratively bend to the magnet from the initial upright position. On the other hand, the initially formed graphene fiber as-released from the pipeline has a porous network structure (Figure S5b,c), which allows to spontaneously immobilize the embedded guest components during drying process. As demonstrated in Figure S10, the TiO₂ nanoparticles have been intercalated into the framework of graphene sheets by soaking the wet graphene fiber in commercial TiO₂ (AEROXIDE P25) aqueous suspension (10 mg/mL) for 20 min with slight stir. After drying and annealing at 400 °C for 30 min, the TiO₂/graphene fiber (~8 wt%) exhibits a fast photocurrent response with good repeatability (Figure 4c). The strong photocurrent response indicates a direct electron/hole injection between the TiO₂ particles and graphene sheets through photoexcitation of TiO₂,[29] suggesting the significant applications in optoelectronic systems such as photodetectors, photocatalysts and photovoltaic cells. In spite of the absence of optimizing the fabrication process in this primary study, Fe₃O₄- and TiO₂/graphene fibers are still flexible and have a half of the tensile strength of neat graphene fibers.
followed by hydrothermal process within pipelines as mentioned above for in-situ incorporation of Fe$_3$O$_4$ nanoparticles into the interlayers of graphene sheets. The dry Fe$_3$O$_4$/graphene fiber with good mechanical flexibility possesses a sensitive magnetic response.

TiO$_2$ nanoparticles (21 nm in diameter) were intercalated into the framework of graphene sheets by soaking the wet graphene fiber in commercial AEROXIDE TiO$_2$ P25 (Degussa Co., Germany) aqueous suspension (10 mg/mL) for 20 min with slight stir. The TiO$_2$ embedded wet graphene fiber (~8 wt%) was dried in air and annealed at 400 °C for 30 min under N$_2$ prior to the photocurrent measurement. The photocurrent response of TiO$_2$/graphene fiber was measured by applying a bias of 0.1 V to the fiber with about 1 cm electrode distance, while the current was recorded by CHI 660D electrochemical workstation upon exposure to a daylight lamp (100 W). More characterization details are included in Supporting Information.

Supporting Information
Supporting Information is available from the Wiley Online Library or from the author.


